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This Is Because Acids And Bases Release H^+ And OH^- ions Respectively In Aqueous Solutions. • Phenolphthalein: Phenolphthalein Remains Colourless In Acids But Turn Pink In Bases. • Methyl Orange: Methyl Orange Turns Pink In Acids And Becomes Yellow In Bases. • Living Organisms Are PH Sensitive. Human Body Works Within A PH Range Of 7.0 To 7.8. Mar 16th, 2019

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• Acids And Bases React To Form Salt And Water. $Acid + Base \rightarrow Salt + H_2O$ • Neutralisation Reaction: Reaction Of Acid With A Base Is Called As Neutralization Reaction. Jan 1th, 2019

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Chapter 23: Acids, Bases, And Salts. Crystalline Solids, Slippery, Strong And Are

Corrosive, Turn Litmus Paper Blue, Does Not Increase The # Of Hydrogen Ions In The Solution., Weak Wren Molecules Break Apart. Jun 5th, 2019

Class10 CHAPTER 2 ACIDS, BASES AND SALTS (Notes)

ACIDS, BASES AND SALTS (Notes) Acids Bases - Sour In Taste - Bitter In Taste - Change The Blue Litmus To Red - Change Red Litmus To Blue - Eg. Hydrochloric Acid HCl Eg. Sodium Hydroxide NaOH - Sulphuric Acid H_2SO_4 Potassium Hydroxide KOH - Nitric Acid HNO_3 Calcium Hydroxide $Ca(OH)_2$ - Acetic Acid CH_3COOH - Ammonium hydroxide NH_4OH Aug 8th, 2019

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Note: This Method Is Used To Make Salts Of Metals Below Hydrogen In The Reactivity Series. If The Base Is A Metal Oxide Or Metal Hydroxide, The Products Will Be Salt And Water Only. If The Base Is A Metal Carbonate, The Products Will Be Salt, Water And Carbon Dioxide. Type 1: Acid + Metal Oxide \rightarrow Salt + Water. To Obtain Copper Sulfate Salt ... Jun 5th, 2019

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Short Notes On Acids (Acids, Basis And Salts) - School

Mineral Acids: Acids Which Are Obtained From Rocks And Minerals Are Called Mineral Acids. Some Common Examples Of Mineral Acids Are Hydrochloric Acid, Sulphuric Acid And Nitric Acid Etc. Note: In General MINERAL Acids Are STRONG Acids While ORGANIC Acids Are WEAK Acids. Jul 22th, 2019

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